

Electromagnetic Radiation

Wave - a traveling disturbance, e.g., displacement of water surface (water waves), string (waves on a string), or position of air molecules (sound waves).

$$h = h_0 \sin\left[\left(\frac{2\pi}{\lambda}\right)(x - vt)\right]$$

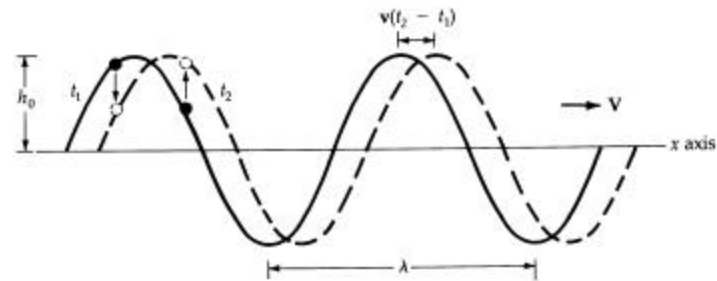
Moving in the $+x$ direction.

Transverse or longitudinal.

h_0 = amplitude; λ = wavelength; v = speed.

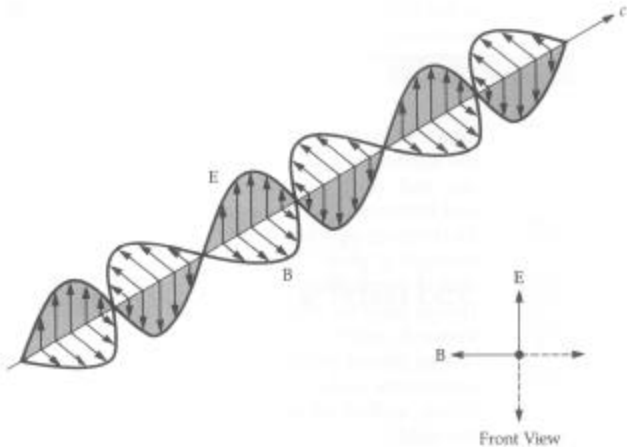
P = period = time for one up-down oscillation at a fixed point = time for wave to move forward one wavelength = λ/v .

Therefore, $v = \lambda \nu$, where $\nu = 1/P$ = frequency.



Electromagnetic Waves

Maxwell (1865): A disturbance in local electric field $\mathbf{E} \Rightarrow$ disturbance in local magnetic field $\mathbf{B} \Rightarrow$ disturbance in \mathbf{E} , etc. Leads to a propagating disturbance (a wave) in a vacuum.



speed of light, $c = 3 \times 10^8$ m/s; $c = \lambda\nu$.

A transverse wave, i.e., disturbance(s) perpendicular to propagation direction.

Source? An oscillating electric charge or current.

Electromagnetic Waves

Consider a wave traveling in the z -direction.

$$E_x = E_{x1} \sin(kz - wt - d_1),$$

$$E_y = E_{y1} \sin(kz - wt - d_2),$$

where $k = 2\pi / \lambda$, $w = 2\pi\nu$, $d = \text{phase}$.

Unpolarized light: \mathbf{E} direction changes randomly; E_x, E_y not correlated; $\delta_1 - \delta_2$ fluctuates randomly.

Plane-polarized light: \mathbf{E} in fixed direction; $\delta_1 - \delta_2 = 0$.

Elliptically polarized light: \mathbf{E} direction changes in a regular way, i.e., \mathbf{E} vector rotates but changes in magnitude. $\delta_1 - \delta_2 = \text{constant}$.

Circularly polarized: \mathbf{E} of fixed magnitude, rotates in a circular path. $E_{x1} = E_{y1}$, $\delta_1 - \delta_2 = \pi/2$.

The Electromagnetic Spectrum

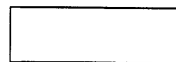
Visible light: $\lambda = 390$ nm (violet) to $\lambda = 720$ nm (far red).

TYPES OF ELECTROMAGNETIC RADIATION

λ	ν or $h\nu$	Type of EM radiation
10^{-5} A	1240 MeV	Gamma rays
10^{-4} A	12.4 MeV	
10^{-3} A	12.4 keV	X-rays
10^{-2} A		
10^{-1} A	124 eV	Ultraviolet
$1\text{A} = 10^{-8}$ cm		
1 nm	1.24 eV	Visible
10 nm		
100 nm	0.012 eV	Infrared
1000 nm = $1\mu\text{m}$		
10 μm	30,000 MHz	Radar
100 μm		
1000 μm	300 MHz	UHF FM
10 mm = 1 cm		
10 cm	3 MHz	Short wave
100 cm = 1 m		
10 m	300 kHz	Broadcast Long Wave
100 m		
1000 m = 1 km	3 kHz	
10 km		
100 km		
1000 km		

Full atmospheric transmission only in visible and some radio wavelengths.

V
B
G
Y
O
R



Open Window



Partial Window



Opaque

eV = electron volt

Hz = hertz = cycles per second

M (mega) = 10^6

k (kilo) = 10^3

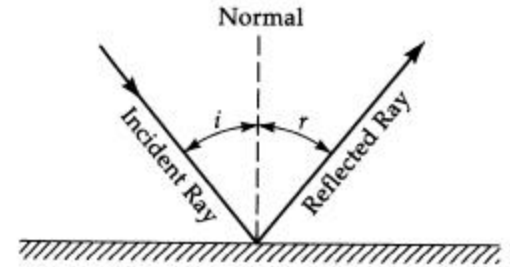
A = angstrom

μm = micrometer

nm = nanometer

Reflection

Use the concept of a ray: a line along the propagation direction of wave crests. Law of reflection: incident angle i of ray equals reflected angle r .



Refraction

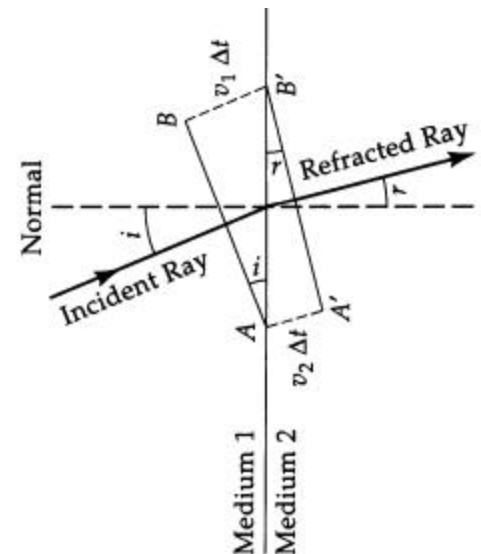
Speed of light decreases when passing through a material medium. Speed equals $v = c/n$, where $n =$ index of refraction. $n = 1.003$ for air, 1.3 for water, about 1.5 for glass.

Light always takes the path of least time.

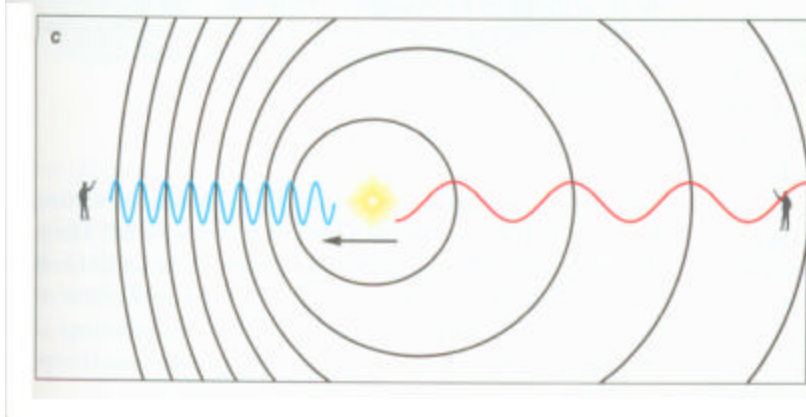
Leads to refraction of wave front when changing media.

$$n_1 \sin i = n_2 \sin r.$$

Telescopes utilize reflection and refraction. In general, $n = n(\lambda)$, which leads to dispersion. Astronomers utilize this for spectroscopy.



The Doppler Effect



Let t be the time between successive crests.

$$l = (c + v)t = c(1 + v/c) / n_0 = l_0(1 + v/c),$$

$$n = c/l = n_0 / (1 + v/c).$$

Note that $v > 0$ if source and observer moving apart, $v < 0$ if moving towards one another.

When v approaches c , need to use relativistic formulas:

$$l = l_0 \left[(1 + v/c) / (1 - v/c) \right]^{1/2},$$

$$n = n_0 \left[(1 - v/c) / (1 + v/c) \right]^{1/2}.$$

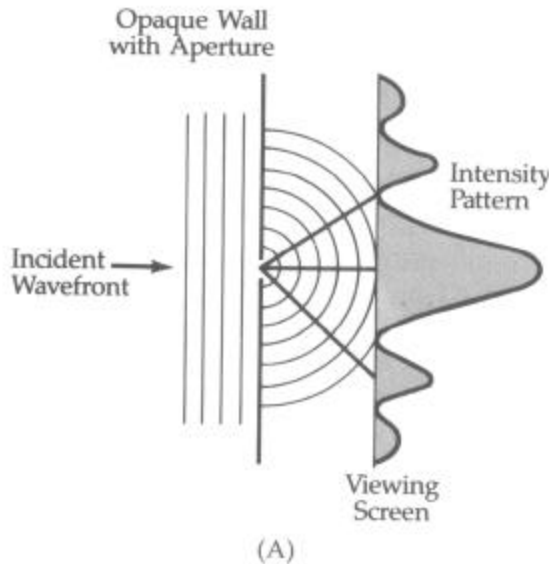
Note convergence to non-relativistic formula for $v \ll c$.

Diffraction and Interference

Electromagnetic radiation has wavelike properties.

We measure the intensity of light $I \propto |E|^2$.

Diffraction - single aperture



Electric field $E = E_1 \sin(kz - \omega t - \mathbf{d})$

Path lengths of waves from different parts of aperture vary. Therefore, waves from two parts of the aperture converging to a single point on the screen yield net field

$$E = E_0 \sin a + E_0 \sin(a + b)$$

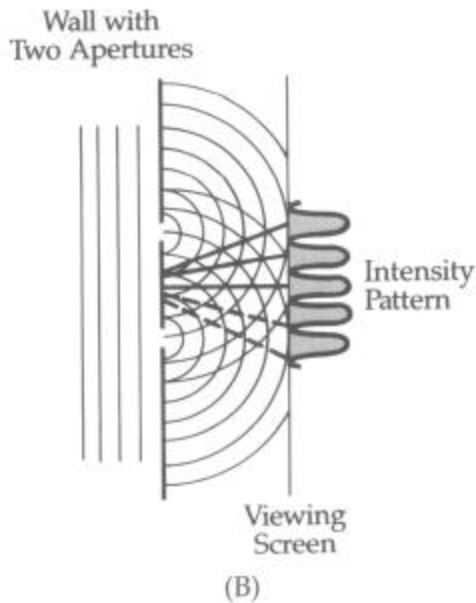
$\Rightarrow I \propto E_0^2 [\sin^2 a + \sin^2 b + 2 \sin a \sin(a + b)]$. Last term can be < 0 .

$I = 0$ when $q \cong \frac{1}{d}$, where d is the diameter of the aperture.

Diffraction is critical for astronomy. Light from a point source is spread over an angle $\theta \cong \frac{\lambda}{d}$, where d is the telescope diameter.

This angle is the resolution limit for a telescope, barring atmospheric turbulence, which limits optical “seeing” to 0.3”-1”.

Interference - multiple apertures



Multiple apertures lead to multiple peaks of width $\theta \cong \frac{\lambda}{a}$, where a is the slit separation.

Since $a \gg d$ in principle, can achieve much higher resolution with multiple receivers => interferometry.

Interferometry



Two views of the Very Large Array (VLA) interferometer at Socorro New Mexico.

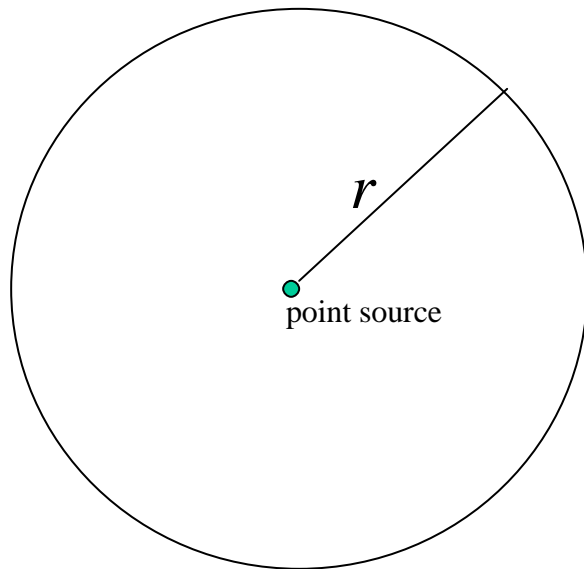


Combined signals from multiple dishes with wide separation yields extremely high resolution, even at (long) radio wavelengths.

Inverse Square Law for Flux of Radiation

A point source of electromagnetic radiation (EMR) emits uniformly in all directions.

Let L be the rate of emission of EMR at all ν (or λ). Units are J/s = W. L is equivalent to the power of the source.



Amount of energy per unit area per unit time received at distance r is the flux

$$F = \frac{L}{4\pi r^2}.$$

Distant objects appear fainter by factor $1/r^2$.

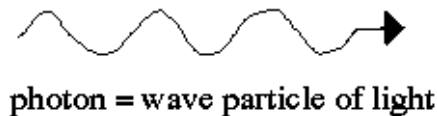
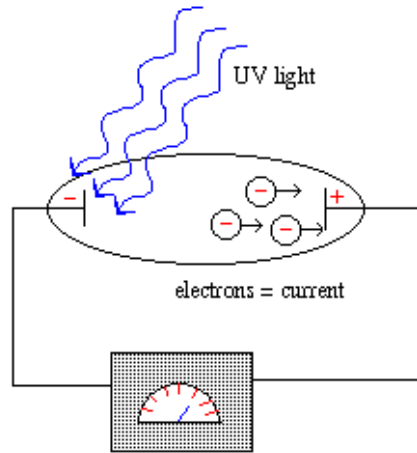
The Quantum Nature of Light

Light has particle-like properties as well!

Picture an EM wave of frequency ν as consisting of many quanta of energy $E = h\nu$, where $h = 6.626 \times 10^{-34}$ J s is Planck's constant.

This model arises from successful theories of blackbody radiation (Planck – 1900) and the photoelectric effect (Einstein – 1905).

Photoelectric Effect



Electron current measured only when ν exceeds a critical value. This is not consistent with classical wave picture, but can be explained if light consists of photons of energy $E = h\nu$.